

Figure 1.

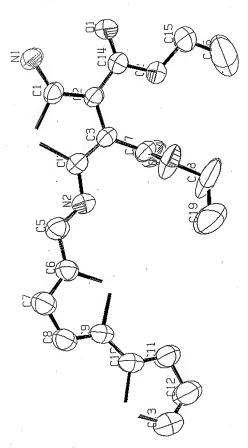


Figure 2.

E _{pc} ² (V)	-1.63	,	-1,84	-1.85	-1.88		1	1	-	Portrod
— ра Г	-1.17	-1.44	-1.12	-1.09	-112	!	ı	,		Dt wire electrode
E _{pa} 3	1	0.83	1.85	1.73			,	1		- chantanta -
$\mathop{\text{\rm S}}_{\text{\rm pa}}^{\text{\rm pa}}$	1.21	0.59	1.02	1 23	1 57		1.12	1 20	1	1
_{га} >	1.43	0.23	0.24	96 0	000	0.30	0.82	0.84	5	
Φ^f (10 ⁻²)	2.3	3.8	0.04	0.5	200	0.42	0.33			
τ (ns)	60	13.5	6.5	9 6	9 1	0.	6.0			
$\overset{E_g}{(\text{eV})^e}$	2,		9 6	3 6	7	6.	25	i	4.4	
ΔΕ (eV) ^d	2.0	2.0	5 0	0.0	7.7	23	27	į	2.5	
λ _{em} (nm) ^c	101	070	2/5	050	610	542	170	ì		
Emax (M ⁻¹ cm	(2	71 820		25 489	. 31530				•	
λ _{abs} (nm) ^b		320	305	440	470	707	1 0	5.4	423	į
Compound		 1	7	m	4	u	> 6	و	7	•

"Scan rate 1V/sec, 0.1 M Bu_kNPF₆, glassy carbon working electron, Ag/AgCl (sat'd) reference electrode, Pt-wire electrode vs. Fe/Fe³t, P_Absorption, "Emission," Refers to absolute HOMO-LUMO difference; "Spectroscopic band-gap; 'Relative to bisthiophene," ^aSpectroscopic band-gap; 'Relative to bisthiophene," ^aLiterature values***

Figure 3.

WO 2005/073265 PCT/CA2005/000131

Compound	Aryl-Aryl ^a	C=Xb	=C-Aryl	Plane Angle ^c		
2	1.443 Å	1.281 Å	1.439 Å	170°		
Analogue ^d	1.479 Å	1.334 Å	1.614 Å	180°		

^abisthiophene distance; ^bX=N for **2** and C for the analogue; ^oRefers to the aryl-C=X dihedral angle; ^dFrom Zobel for bisthiophene and thiophene alkene values. ¹²

Figure 4.

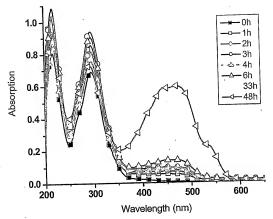


Figure 5

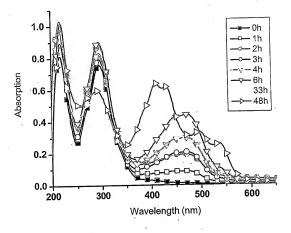


Figure 6.

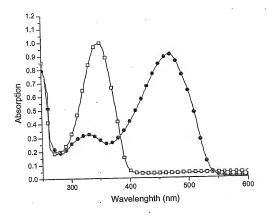


Figure 7.

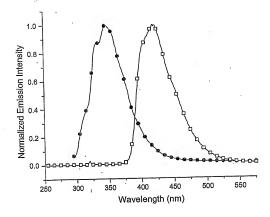


Figure 8.

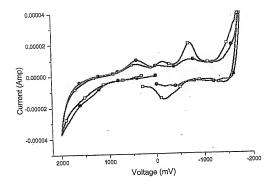


Figure 9.

WO 2005/073265 PCT/CA2005/000131

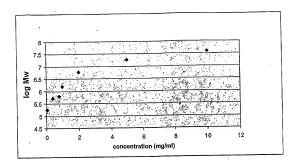


Figure 10.